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FLAVONOIDS FROM *ABRUS PRECATORIUS*

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(Received 4 January 1980)

Key Word Index—*Abrus precatorius*; Leguminosae; flavonoids; 6,4'-dimethoxy-7,3'-dihydroxyflavone; abrectorin; desmethoxycentaureidin 7-O-rutinoside.

Abrus precatorius, having medicinal properties [1], has been examined for its chemical components [2–6]. This communication describes the isolation of a new flavone, abrectorin and a known glycoside, desmethoxycentaureidin 7-O-rutinoside.

A. precatorius seed kernels (2 kg) were defatted using petrol and then extracted exhaustively with EtOH. Solvent-free EtOH extract was repeatedly extracted with Et₂O and then EtOAc to separate non-glycosidic and glycosidic components. Combined Et₂O and EtOAc extracts containing non-glycosidic components were concentrated. The solvent-free concentrate was chromatographed on a Si gel column; elutions with C₆H₆–EtOAc (3:1) gave a mixture which was separated into two compounds by preparative TLC (Si gel; C₆H₆–EtOAc; 1:1). One of these was characterized as luteolin by direct comparison with an authentic sample whereas the second compound (**1**) was a new flavone, here named abrectorin. Et₂O- and EtOAc-insoluble fractions containing glycosidic components were combined, concentrated and the resulting solvent-free concentrate was chromatographed on a Si gel column, which was eluted with EtOAc–MeOH (3:1) to give a mixture of three compounds which on preparative-PC using *n*-BuOH–HOAc–H₂O (4:1:5; upper layer) yielded orientin, isoorientin and a flavone glycoside (**2**).

Abrectorin (**1**)

Colour reactions and spectral data indicated **1** to be a polyhydroxyflavone. Methylation of **1** yielded a dimethyl

ether identical with 6,7,3',4'-tetramethoxyflavone (**1a**) [Bhardwaj, D. K. *et al.*, unpublished results] showing that **1** was a 6,7,3',4'-dihydroxydimethoxyflavone. It gave a negative Quastel test [7] showing the absence of an *ortho*-dihydroxyl in the molecule. On alkali fission **1** yielded isovanillic acid, fixing an OH and an OMe at C-3' and C-4' respectively. The solubility of **1** in aqueous Na₂CO₃ (10%) indicated another OH at C-7, so that the other OMe was therefore at C-6. Abrectorin is thus 6,4'-dimethoxy-7,3'-dihydroxyflavone (**1**). This was confirmed by the identity of its diethyl ether with synthetic 6,4'-dimethoxy-7,3'-diethoxyflavone (**1b**) [8].

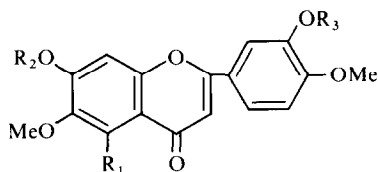
Desmethoxycentaureidin 7-O-rutinoside (**2**)

Colour reactions and spectral data indicated **2** to be a polyhydroxyflavone glycoside. On hydrolysis it yielded an aglycone (**2a**) and two free sugars identified as rhamnose and glucose by PC (*n*-BuOH–Py–H₂O, 6:4:3). Methylation of **2a** yielded a trimethyl ether identical with 5,6,7,3',4'-pentamethoxyflavone (**2b**) [9] indicating that **2a** was a 5,6,7,3',4'-trihydroxydimethoxyflavone. On alkali fission **2a** yielded isovanillic acid showing a OH and a OMe at C-3' and C-4' respectively. **2a** did not give the Bargellini test for a 5,6,7-trihydroxy system or the Quastel test [7] characteristic for an *ortho*-dihydroxy system. Consequently, the remaining OMe and OH were at C-6 and C-7 respectively. Bathochromic shifts in UV spectrum of **2a** with AlCl₃ confirmed the chelated OH at C-5 and its solubility in aqueous Na₂CO₃ (10%) supported another OH at C-7. The absence of the usual bathochromic shifts in

UV spectrum of **2a** with NaOAc as generally recorded for the 7-hydroxyflavones was attributed to the presence of OMe at C-6 position [9]. The aglycone is thus 5,7,3'-trihydroxy-6,4'-dimethoxyflavone (desmethoxycentaureidin) [10]. Acetylation of **2** gave an acetate **2c**. $^1\text{H NMR}$ of **2c** showed two OMe, two phenolic COMe, six alcoholic COMe, sugar protons besides those for five aromatic protons. **2** on complete methylation followed by hydrolysis gave a methyl ether soluble in aqueous Na_2CO_3 (10%) and also identical with 5,6,3',4'-tetramethoxy-7-hydroxyflavone (**2d**) [9]. Hence, **2** is the 7-*O*-diglycoside of **2a**. Permethylation followed by hydrolysis of **2** gave 2,3,4-tri-*O*-methyl-D-glucopyranose (R_G 0.84) and 2,3,4-tri-*O*-methyl-L-rhamnopyranose (R_G 1.02) indicating it to be a rutinoid [11] as was confirmed by the $^1\text{H NMR}$ of its acetate (**2c**) [12]. The glycoside is thus the known desmethoxycentaureidin 7-*O*-rutinoside (**2**) [13].

EXPERIMENTAL

1: Creamish micro-prisms from MeOH; mp 229–230° (Found: C, 65.2; H, 4.8. $\text{C}_{17}\text{H}_{14}\text{O}_6$; requires: C, 64.96; H, 4.49%); developed yellow–orange colouration with Mg-HCl ; was soluble



- 1** $R_1 = R_2 = R_3 = \text{H}$
1a $R_1 = \text{H}; R_2 = R_3 = \text{Me}$
1b $R_1 = \text{H}; R_2 = R_3 = \text{Et}$
2 $R_1 = \text{OH}; R_2 = \text{rutinosyl}, R_3 = \text{H}$
2a $R_1 = \text{OH}; R_2 = R_3 = \text{H}$
2b $R_1 = \text{OMe}; R_2 = R_3 = \text{Me}$
2c $R_1 = \text{OAc}; R_2 = \text{hexa-}O\text{-acetyl rutinosyl}; R_3 = \text{Ac}$
2d $R_1 = \text{OMe}; R_2 = \text{H}; R_3 = \text{Me}$

in aq. Na_2CO_3 (10%); IR (KBr) cm^{-1} : 3448 (OH), 1667 (conj. CO); UV (MeOH) nm: 250, 290, 335. No UV spectral shifts with NaOAc. **1a**: Methylation of **1** with Me_2SO_4 (2.2 mol)/ K_2CO_3 in Me_2CO gave **1a**, colourless needles from CHCl_3 –petrol; mp 221–222° (Found: C, 66.40; H, 5.60. $\text{C}_{19}\text{H}_{18}\text{O}_6$ requires: C, 66.66; H, 5.30%); $^1\text{H NMR}$ (CDCl_3): δ 3.99 (12 H, s, $4 \times -\text{OMe}$), 6.60 (1 H, s, C-3-H), 6.87 (1 H, s, C-8-H), 7.02 (1 H, d, $J = 9$ Hz, C-5'-H), 7.45 (2 H, m, C-2'-H and C-6'-H), 7.52 (1 H, s, C-5-H). **1b**: Ethylation of **1** with Et_2SO_4 (2.2 mol)– K_2CO_3 in Me_2CO gave **1b**, creamish needles from MeOH; mp 176°.

2: Pale-yellow needles from MeOH–EtOAc, mp 183–184°; IR (KBr) cm^{-1} : 3450 (OH), 1640 (conj. CO); UV (MeOH) nm: 275, 335; + AlCl_3 280, 305, 360. **2a**: Hydrolysis of **2** with MeOH– H_2SO_4 (7%) gave rhamnose and glucose as free sugars and an aglycone **2a**, yellow needles from EtOAc, mp 269–270°

(Found: C, 62.3; H, 4.6. $\text{C}_{17}\text{H}_{14}\text{O}_7$ requires: C, 61.82; H, 4.24%); UV (MeOH) nm: 275, 335; + AlCl_3 280, 305, 355. **2a** on complete methylation with Me_2SO_4 – K_2CO_3 in Me_2CO gave **2b** [9], rectangular plates from EtOH, mp 178°. **2c**: Acetylation (Ac_2O –Py) of **2** gave **2c**, colourless needles from EtOAc–petrol, mp 121–122°. $^1\text{H NMR}$ (CDCl_3): δ 1.18 (3 H, br s, rhamnosyl C-Me), 1.80–2.05 (18 H, m, $6 \times -\text{OCOMe}$, alcoholic), 2.34 (3 H, s, $1 \times -\text{OCOMe}$, phenolic), 2.47 (3 H, s, $1 \times -\text{OCOMe}$, phenolic), 3.90 (3 H, s, $1 \times -\text{OMe}$), 4.10 (3 H, s, $1 \times -\text{OMe}$), 3.47–5.28 (12 H, m, sugar-protons), 6.50 (1 H, s, C-3-H), 6.70 (1 H, s, C-8-H), 7.32 (1 H, d, $J = 9$ Hz, C-5'-H), 7.42 (1 H, d, $J = 2$ Hz, C-2'-H), 7.77 (1 H, m, C-6'-H). **2d**: Methylation of **2** with Me_2SO_4 – K_2CO_3 in Me_2CO , followed by hydrolysis with MeOH– H_2SO_4 (7%) gave a methyl ether **2d**, pale-yellow needles from MeOH, mp 220–222°, agreed with 5,6,3',4'-tetramethoxy-7-hydroxyflavone (**2d**) [10].

Permethylation and hydrolysis of 2. **2** (8 mg) in DMSO (2 ml) and NaH dispersion in oil (50%, 10 mg) was left at 80° for 1 hr, cooled, treated with MeI (1 ml), left overnight, poured into ice-cold H_2O and then extracted with CHCl_3 . The permethylated product was hydrolysed using Killiani mixture (3 ml). The hydrolysate was found to contain (2,3,4-tri-*O*-methyl-D-glucopyranose (R_G 0.84) and 2,3,4-tri-*O*-methyl-L-rhamnopyranose (R_G 1.02) as shown by direct comparison with authentic samples by PC in *n*-BuOH–EtOH– H_2O (5:1:4).

Acknowledgements—We wish to thank the University Grants Commission, and Council of Scientific and Industrial Research, New Delhi, for the financial grants and Dr. R. Singh for helpful discussions.

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